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CADMIUM- IMPACTS ON PLANT AND PHYTOREMEDIATION : A REVIEW

Namrata Maity¹ and Manabesh Majumdar^{2*}

1. Department of Botany, The University of Burdwan, PIN-713104, West Bengal, India Email : namratamaity1997@gmail.com

2. Department of Zoology, Bejoy Narayan Mahavidyalaya, Itachuna, PIN-712147, Hooghly,

West Bengal, India

Email: manabesh@bnmv.ac.in

* Corresponding author

Abstract

Today a major concern is heavy metal toxicity. Cadmium is non-essential, non-biodegradable highly toxic and may act as a carcinogen. Along with natural sources; several anthropogenic activities increase its toxicity in environment. It shows its toxicity at very low concentration. Cadmium can translocate to shoot and another plant parts through root system. It has bad impact on plants, microorganisms, and also on human beings. Cadmium has a role on seed germination, growth and development, photosynthesis, ROS production etc. To combat its bad effects, plants have employed several phytoremediation processes.

1. Introduction

Now a days, a long-lasting and increasing threat is heavy metal toxicity to environment. Most of the heavy metals are trace elements in plant having significant role in several physiological functions at low concentration. Due to presence of high conc. of them, soil becomes polluted. Sources of heavy metals are coal combustion, expanding industrial emissions, fertilizers, pesticides, petrochemical spillage, sewage sludge etc. [1]. Even at low concentration it shows

toxicity [2, 3]. Cadmium (Cd), mercury (Hg), lead (Pb), nickel (Ni), zinc (Zn), iron (Fe), manganese (Mn), magnesium (Mg), cobalt (Co), copper (Cu), antimony (Sb), chromium (Cr) etc. are very commonly present in soil [4, 5, 6]. In soil maximum inorganic pollutants don't undergo microbial or chemical degradation; they persist as their own form for a long time [7].

Highly toxic, non-essential, non-biodegradable, carcinogenic cadmium is electropositive, soft, silver white metal placed 'd' block and ' Group 12' in periodic table. It has more than 20years of biological half-life with atomic number 48, atomic mass 112.41, density 8.64 g/cm³ and melting temperature 321° C [8, 9].

It is the general consideration that if concentration of the total cadmium (Cd) in soil exceeds 8 mg kg $^{-1}$, or its bioavailability becomes >0.001 mg kg $^{-1}$, or its concentration reaches 3–30 mg kg $^{-1}$ in plant tissues, most plants exhibit visible Cd toxicity symptoms [10].

Pure cadmium can't be present in soil. It remains as ores of carbonate, sulphides and oxides of lead, zinc, copper etc. Oxides are less water soluble than sulphide and carbonate which are toxic to plants [11, 12]. Through root system, it can translocate into plant parts. Not being only restricted in root system, it also becomes transported to shoot and accumulate in edible part. P-type (P1B) ATPase is responsible for cadmium translocation from root too shoot. e.g, OsHMA3, a heavy metal ATPase 3 and OsHMA2 are found in case of rice plant [13, 14].

Easily identifiable symptoms are chlorosis and stunted growth in plants [15]. Several scientific studies on cadmium toxicity have impacts on ROS production, crop productivity, lipid peroxidation etc. as well as remediation of all these problems [16, 17, 18, 19, 20].

2. Roles of cadmium on plant

- a) On seed germination:
- Increase of Cd²⁺conc. at μmol/L level has inhibitory/ reducing effects on seed germination in soybeans, lettuce and sugar beet and spinach [21].
- In case of *Vigna unguiculata* L. seed, cadmium decreases the water absorption. Thus, water supply is limited for seed embryo development [22].
- Due to cadmium stress, α-amylase activity is reduced which decreases starch release from cotyledons [23].
- Calmodulin and cadmium relationship has an important role in metabolic activation during early stages of seed germination. In *Raphanus sativus L*. Ca and Cd shows competition for Ca-calmodulin binding sites [24].

b) On plant growth, development and dry weight:

- Cadmium stress hampers root elongation in crop plants like rice, wheat, tomato etc. [25, 26, 27].
- After long-term Cd exposure, root elongation reduces in crop plants because of decomposition, necrosis and mucilaginous deposition and shoots also become reduces. As a result, leaf rolling and chlorosis occur [25, 26, 27].
- Cadmium toxicity responsible for common abnormalities in roots like fragmentation, stickiness, precocious separation, bridges and ligands [28]
- Accumulation of high amount of Cd^{2+} , plant root length decreases. e.g.; 0.02 cm is reduced due to 70 μ mol/L Cd treatment. Shoot length also becomes suppressed for increasing cadmium conc.
- Because of abnormal enlargement of cortical cell layers and apical part of epidermis, tap root becomes rigid, brown and twisted [29, 30]. Cadmium is also responsible for inhibiting lateral root development and tap root development [29].
- By minimizing mitotic division of meristematic cells, cadmium can reduce root length, dry biomass and increases root diameter [31, 32].
- Total leaf area and dry weight of plant decrease under Cd stress [33].

c) On photosynthesis:

- Cadmium is an effective photosynthetic inhibitor [34]. In leaves, Cd accumulation inhibits the stomata opening in oilseed, legume and cereal crops. This indicates a relationship between transpiration and inhibition of photosynthesis [35].
- Short-term and long-term exposure to cadmium toxicity hampers the photosynthetic activity in many crops including rapeseed [36], pea, maize, barley [37], wheat [27] and mungbean [38].
- Light harvesting system and Photosystem I and II are affected by Cd toxicity [39].
- Primary action sites of cadmium are photosynthetic apparatus, pigments, chlorophyll and carotenoids synthesis [40]. Cadmium has ability to minimize chloroplast density which interacts with chlorophyll biosynthesis. As a result, chlorosis occurs in oilseed crop [41]. This chlorophyll amount reduces more abundantly in stomatal guard cell than mesophyll cell.
- Rubisco and PEPCase enzymes play an important role in carbon fixation during photosynthesis. By replacing co-factor Mg²⁺ ion, responsible for carboxylation reaction, Cd²⁺ decreases Rubisco activity. Replacement of Mg ion shifts carboxylation activity of Rubisco towards oxygenation reaction [37].

d) On Protein and amino acids:

- Under stress condition, eukaryotes can synthesize heat-shock proteins [15]. 70 kDa phosphoprotein, a heat-shock protein is synthesized by maize under Cd stress condition.
- In Pea heat-shock protein HSP71 and pathogen-related protein-PrP4A have important role to protect the cell from Cd toxicity [42].

e) On plant-water relation:

• Water-related changes occur across the entire plant because of cadmium exposure. In roots, water absorption is reduced and short distance water transport in apoplast and symplast pathways are inhibited by Cd toxicity.

• According to Malecka et al., 2008, cadmium enriched soil has lower osmotic ability than root cell sap. This condition can limit plant water adsorption and causes osmotic pressure [43].

f) On ROS production

- Due to Cd exposure reactive oxygen species (ROS) production is induced and toxic effects results from this oxidative stress in a cascade manner.
- Cd can play the role of a co-factor by replacing the essential metalloprotein ions, the so called Fanton reaction [44, 45]. As a result of this alter reaction, lipid peroxidation becomes induced in plant cell due to the production of substances like ROS, O_2^- , OH and H_2O_2 and all these happens for essential nutrient uptake [46, 47]. These results cell membrane destruction.
- Even ion and redox homeostasis also get disturbed in cellular metabolic activities [48]. Basically over production of O₂⁻ takes place as a result of Cd induced oxidative stress in parts like root, shoot etc. in seedlings.

3. Remediation of cadmium toxicity

I. Cadmium tolerance capacity of plant:

- **Plant species:** Different plant species like beet, turnips and kale family have different levels of sensitivity to different trace elements and are much sensitive to metals. Soybeans, spinach, lettuce etc. are moderately sensitive to soils, containing cadmium [49]. In high cadmium containing soils (i.e., >100ppm), rice, tomato, cabbage, squash etc. can grow.
- Age and plant species: In Quack grass, early spring shoots can accumulate high conc. of trace elements like Cd, Cu, Cr, Hg, Pb, Zn, Ni and Mn whereas tissues at later stage can accumulate these elements in low conc. [50]. Therefore, it is sure that most of the consumable parts do not contain the significant level of heavy metal.

Different plant parts contain different conc. of trace elements such as vegetative tissue contains large amount of metals than seeds or grains. In sludge treated soils, growing corn accumulate Cd and Zn conc. in following order- leaf> stem> husk> kernel.

II. Phytoremediation: In in-situ bioremediation approach, direct use to accumulate, extract, sequester and to immobilize or detoxify cadmium in soil is known as phytoremediation of cadmium [51]. It is cost-effective and eco-friendly. For treating large, diffused contaminated area, phytoremediation is used while physical and chemical remediation technique is strictly used in small contaminated area. Plants, those contribute for phytoremediation, possess rapid growth, high biomass production, deep and dense root system and high coefficient of bioaccumulation [52]. Phytoremediation is classified into different categories like phytostabilization/ phytoimmobilization/ phytorestoration, phyto-extraction/phytoaccumulation, rhizofiltration, phyto-stimulation, phyto-volatilization/ rhizovolatilization and phytodegradation [53, 54]. Phytoextraction is the prime process used for Cd remediation [55, 56]. Brief description of these processes are as follows which are applicable for Cd remediation-

Phyto-extraction/ Phyto-accumulation: It is a low impact technology. To remove Cd from contaminated soil or water at low and high concentration, phyto-extraction mechanism is involved [57]. Based on toxin translocation from root to shoot and leaves, this mechanism is established [58]. Along with nutrient and water, plants accumulate contaminants from contaminated place. They can't destroy these contaminants but prefer to accumulate in shoots, leaves and other parts of plants [59]. To neutralize the effect of metallic and 'radioactive' species waste plants widely utilized the phytoextraction process which has a great scope to establish the commercialization of this method having low initial investments and great opportunity to mitigate environmental problem of those lands contaminated by heavy metals [60]. Phyto-extracts of rapeseed can reduce 60% Cd in soil than control [61]. Plants used in phyto-extraction, possess following characters- (i) have ability for accumulating high conc. of heavy metals within their biomass, (ii) capacity of high absorption as well as translocation, (iii) high growth rate and extensive root system, (iv) high tolerance to contaminants. For determining plant potential of accumulation of heavy metals, several aquatic plant species like Myriophyllum aquaticum [62], Mentha aquatic [63], Pistia stratiotes, Spirodela polyrrhiza, Ludwigina palustris [64] were studied. High amounts of Cd can be accumulated by grain crops. Some root crops and leafy crops can able to accumulate cadmium [65]. Table 1 shows some cadmium accumulating plant species.

Name of plant species	Cadmium accumulating medium	Cadmium accumulating plant part	References
Solanum photeincarpum	soil	Root, stem	[66]
Brassica junea	soil	-	[67]
Thlaspicaerulescens	soil	Shoots	[68]
<i>Arabis paniculata</i> Franch	water	Roots	[69]
Cyperus rotundus	-	-	[70]
Amaranthus hypochondriacus L.,	soil	-	[71]
Solanum nigrum L.,	soil	-	[71]

Table 1: Cadmium accumulating plant species:

Phytolacca acinosa Roxb.,	soil	-	[71]
Celosia argentea L.,	soil	-	[71]

Several chemicals or surfactants viz. EDTA (Ethylenediamine tetraacetic acid) [72, 73], CDTA [74], DTPA, EDDHA [75], citric acid [72] and NTA [76] can increase mobility of metals in soil. In plants the absorption rate of metals is also enhanced by employing these chemicals [77].

Phytostabilization / phytoimmobilization: In this process, contaminants are immobilized to other plant parts and restricted at root region. Plants utilized in this process can accumulate heavy metals by root hair, root surface adsorption or precipitation by rhizosphere [78, 79, 80]. This process reduces the bioavailability of heavy metals and prevents entering it to food chain. By this process, soil metal transform into less toxic form [81]. In case of high organic content containing soil, this process is very effective [82]. This method is helpful to rectify the contaminated sites having high metal concentration to re-instead the proper vegetation. Some metal tolerant plant species can be utilized to control the emigration of different contaminants by agents like rain, wind and leaching into groundwater [83]. Plant associated microbiota assists in this process. Microbe- associated heavy metal resistance process are as follows- (i) by permeability barrier/ active expulsion of metal outside the cell, (ii) by using extracellular polymer, (iii)to transform into less toxic form of metal [84]. In phytostabilization, combinedly grass and trees work better due to non-accumulation of metals by grass in their shoots. This process reduces toxic metal exposure to animals [85]. Table 2 shows cadmium phytostabilizing plant species.

Table 2

Name of the plant species	References
Tamarindus Indica	[86]
Populus cathayana, P. prezwaskii, P. yunnanensis	[57]
Nicotiana tabacum L.	[87]
Zea mays	[88]
Sorghum	[89]

Cadmium phytostebilizing plant species.

* *Rhizofiltration:* This method works with the synthesis of several chemicals in the roots to adsorb pollutants as plants of some nature may contain several phytochelating to bind with metal ion pollutants and thus increase its adsorption [90]. To retain Pb, Cd, Cu Zn, Ni and Cr within root, rhizofiltration is needed [91]. For remediation of waste water, ground water and surface water containing low conc. of contaminant, this technique is used [92, 93]. In this mechanism the plant utilized, possess hairy and longer root system with considerable surface area for filtering the pollutant from aqueous solutions [94]. Table 3 shows rhizofiltration plant species.

rhizofiltration plant species				
SI. No.	Name of plant species	References		
1.	Brassica napus	[95]		
2.	Helianthus annuus L	[95]		

Table-3

Above-mentioned two plants species are favorable for rhizofiltration process because they possess the aforesaid characters.

* *Phyto-volatilization*: According to Moreno et al., 2004 by using plant, uptake of contaminants, increasing and release of these contaminants in environment as less harmful form is known as phyto-volatilization [96]. These less toxic volatile materials are formed by using plant-based transpiration cycle [97]. Because of not're-deposition' at/ or near the site of gaseous volatilized products phyto-volatilization considered as permanent contaminated site solution. Table 4 shows cadmium volatilizing plant species.

Table-4

Cadmium volatilizing plant species

SI. No.	Name of plant species	References
1.	Silphium perfoliatum	[98]
2.	Acanthus ilicifoliusL.	[99]
3.	Typha latifolia L.	[100]
4.	Cynodon dactylon	[100]
5.	Quercus ilex	[101]

Rhizodegradation: is a plant species-microbes specific symbiotic process [102, 103]. Root associated microbes can remediate contaminated soil by rhizodegradation, a rhizospheric biological degradation process. Plants supply energy sources i.e., carbon compounds to associated microbes to enhance the metabolic activities for degradation or removal of cadmium from contaminated soil. Edaphic and environmental factors have great impact on this symbiotic relationship. e.g.; soil pH can change Cd bioavailability in plant associated rhizospheric plane. Cd translocation and bioaccumulation are significantly increased by oxalic acid and citric acid of root exudates of *Echinochloa crusgalli* [104]. By changing several physical, chemical and biological conditions like pH, moisture, temperature, organic matter, microbial metabolism etc. we can enhance or suppress rhizodegredation process.

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References

[1] M. K. Zhang, , Z. Y. Liu, & H. Wang, "Use of single extraction methods to predict bioavailability of heavy metals in polluted soils to rice", Soil science and plant analysis, 41(7), 820-831 (2010).

[2] P. Babula, , V. Adam, , R. Opatrilova, , J. Zehnalek, , L. Havel, , &R. Kizek, "Uncommon heavy metals, metalloids and their plant toxicity: a review", Organic Farming, Pest Control and Remediation of Soil Pollutants, 275-317 (2009).

[3] A. Bhargava, F. F. Carmona, M. Bhargava, & S. Srivastava, "Approaches for enhanced phytoextraction of heavy metals", Journal of environmental management, 105, 103-120 (2012).

[4] O. E. Orisakwe, "Other heavy metals: antimony, cadmium, chromium and mercury. In Toxicity of building materials Woodhead Publishing, 297-333 (2012).

[5] R. A. Street, "Heavy metals in medicinal plant products—An African perspective", South African Journal of Botany, 82, 67-74 (2012).

[6] L. H. Allen, , Food Safety: Heavy Metals, third ed. Encyclopedia of Human nutrition. 331-336 (2014).

[7] A. Kubier, R. T. Wilkin, , & T. Pichler, "Cadmium in soils and groundwater: a review", Applied Geochemistry, 108, 104388 (2019).

[8] E. Aksoy, J. Salazar, &H. Koiwa, "Cadmium determinant 1 is a putative heavy-metal transporter in Arabidopsis thaliana (617.4)" *The FASEB Journal* 28, 617-4 (2014).

[9] N Vinodini, PK Chatterjee, P Chatterjee, S Chakraborti, A Nayanatara, RM Bhat, K Rashmi, V Suman, SB Shetty, SR Pai, "Protective role of aqueous leaf extract of Moringa oleiferaon blood parameters in cadmium exposed adult wistar albino rats", Inter J Curr Res Acad Rev. 2015;3:192–9 (2015).

[10]A. Dutta, A. Patra, H. S. Jatav, S. S. Jatav, S. K. Singh, E. Sathyanarayana, S. Verma, & P. Singh, "Toxicity of Cadmium in Soil-Plant-Human Continuum and Its Bioremediation Techniques" M. L. Larramendy, & S. Soloneski (Eds.), Soil Contamination - Threats and Sustainable Solutions (2020).

[11] M. Monachese, J. P. Burton, & G. Reid, "Bioremediation and tolerance of humans to heavy metals through microbial processes: a potential role for probiotics?", Applied and environmental microbiology, 78(18), 6397-6404 (2012).

[12] A. M. Darwish, W. H. Eisa, A. A. Shabaka, & M. H. Talaat, "Investigation of factors affecting the synthesis of nano-cadmium sulfide by pulsed laser ablation in liquid environment". Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 153, 315-320 (2016).

[13] H. Miyadate, S. Adachi, A. Hiraizumi, K. Tezuka, N. Nakazawa, T. Kawamoto, K. Katou, I. Kodama, K. Sakurai, H. Takahashi, N. Satoh-Nagasawa, A. Watanabe, T. Fujimura, & H. Akagi, "OsHMA3, a P1B-type of ATPase affects root-to-shoot cadmium translocation in rice by mediating efflux into vacuoles", The New phytologist, 189(1), 190–199 (2011).

[14] R. Takahashi, Y. Ishimaru, H. Shimo, Y. Ogo, T. Senoura, N. K. Nishizawa, &H. Nakanishi, "The OsHMA2 transporter is involved in root-to-shoot translocation of Zn and Cd in rice", Plant, cell & environment, 35(11), 1948–1957 (2012).

[15] P. Jali,, C. Pradhan, , & A. B. Das, "Effects of cadmium toxicity in plants: a review article", Scholars Academic Journal of Biosciences (SAJB), 4(12), 1074-1081 (2016).

[16] S. Bashir, M. S. Rizwan, A. Salam, Q. Fu, J. Zhu, M. Shaaban, & H. Hu, "Cadmium immobilization potential of rice straw-derived biochar, zeolite and rock phosphate: extraction techniques and adsorption mechanism", Bulletin of environmental contamination and toxicology, 100(5), 727-732 (2018).

[17] N. Baruah, , S. C. Mondal, M. Farooq, &N. Gogoi, "Influence of heavy metals on seed germination and seedling growth of wheat, pea, and tomato", Water, Air, & Soil pollution, 230(12), 1-15 (2019).

[18] S. QIN, H. LIU, , & Z. NIE, Z. Rengel, GAO Wei, LI Chang, Z. Peng, "Toxicity of Cadmium and Its Competition with Mineral Nutrients for Uptake by Plants: A Review", Pedosphere, 30, 168-180 (2020).

[19] B. Hussain, M. N. Ashraf, A. Abbas, J. Li, & M. Farooq, "Cadmium stress in paddy fields: effects of soil conditions and remediation strategies" Science of The Total Environment, 754, 142188 (2021).

[20] B. Hussain, M. J. Umer, J. Li, Y. Ma, Y. Abbas, M. N. Ashraf, & M. Farooq, "Strategies for reducing cadmium accumulation in rice grains", Journal of Cleaner Production, 286, 125557 (2021).

[21] O. V. Bautista, Fischer, G., & Cárdenas, J. F. "Cadmium and chromium effects on seed germination and root elongation in lettuce, spinach and Swiss chard", Agronomía colombiana, 31(1), 48-57 (2013).

[22]D. Thamayanthi, P. S. Sharavanan, &M. Vijayaragavan, "Effect of cadmium on seed germination, growth and pigments content of Zinnia plant" *Current Botany*, 2(8), 8-13 (2011).

[23]T. Kalai, D.Bouthour, J. Manai, L. Bettaieb Ben Kaab, &H. Gouia, "Salicylic acid alleviates the toxicity of cadmium on seedling growth, amylases and phosphatases activity in germinating barley seeds", *Archives of Agronomy and Soil Science* 62(6), 892-904 (2016).

[24] M. Huybrechts, A. Cuypers, J. Deckers, V. Iven, S. Vandionant, M. Jozefczak, & S. Hendrix, "Cadmium and plant development: an agony from seed to seed", *International journal of molecular sciences* 20(16), 3971 (2019).

[25] M. Rizwan, J. D. Meunier, H. Miche, & C. Keller, "Effect of silicon on reducing cadmium toxicity in durum wheat (Triticum turgidum L. cv. Claudio W.) grown in a soil with aged contamination" *Journal of hazardous materials* 209, 326-334 (2012).

[26] I. Hussain, , M. A. Ashraf, R. Rasheed, A. Asghar, M. A. Sajid, & M. Iqbal, "Exogenous application of silicon at the boot stage decreases accumulation of cadmium in wheat (Triticum aestivum L.) grains" *Brazilian Journal of Botany* 38(2), 223-234 (2015).

[27] T. Abbas, M. Rizwan, S. Ali, M. Zia-ur-Rehman, M. F. Qayyum, F. Abbas, & Y. S. Ok, "Effect of biochar on cadmium bioavailability and uptake in wheat (Triticum aestivum L.) grown in a soil with aged contamination" Ecotoxicology and Environmental Safety, 140, 37-47 (2017).

[28] X. M. Liu, K. E. Kim, K. C. Kim, X. C. Nguyen, H. J. Han, M. S. Jung, & W. S. Chung, "Cadmium activates Arabidopsis MPK3 and MPK6 via accumulation of reactive oxygen species" *Phytochemistry* 71(5-6), 614-618 (2010).

[29]A. Krantev, R. Yordanova, T. Janda, G. Szalai, & L. Popova, "Treatment with salicylic acid decreases the effect of cadmium on photosynthesis in maize plants" *Journal of plant physiology* 165(9), 920-931 (2008).

[30] N. Rascio, & F. Navari-Izzo, "Heavy metal hyperaccumulating plants: how and why do they do it? And what makes them so interesting?" *Plant science* 180(2), 169-181 (2011).

[31] C. S. Seth, V. Misra, L. K. S. Chauhan, & R. R. Singh, "Genotoxicity of cadmium on root meristem cells of Allium cepa: cytogenetic and Comet assay approach" Ecotoxicology and Environmental safety 71(3), 711-716 (2008).

[32]P. L. Gratão, C. C. Monteiro, M. L. Rossi, A. P. Martinelli, L. E. Peres, L. O. Medici, & R. A. Azevedo, "Differential ultrastructural changes in tomato hormonal mutants exposed to cadmium", Environmental and Experimental Botany 67(2), 387-394 (2009).

[33] N. Jinadasa, D. Collins, P. Holford, P. J. Milham, & J. P. Conroy, "Reactions to cadmium stress in a cadmium-tolerant variety of cabbage (Brassica oleracea L.): is cadmium tolerance necessarily desirable in food crops?" *Environmental Science and Pollution Research* 23(6), 5296-5306 (2016).

[34]A. Vassilev, A. Perez-Sanz, B. Semane, R. Carleer, & J. Vangronsveld, "Cadmium accumulation and tolerance of two Salix genotypes hydroponically grown in presence of cadmium" *Journal of plant nutrition* 28(12), 2159-2177 (2005).

[35] F. Zhang M. Liu, Y. Li, Y. Che, & Y. Xiao, "Effects of arbuscular mycorrhizal fungi, biochar and cadmium on the yield and element uptake of Medicago sativa" Science of The Total Environment 655, 1150-1158 (2019).

[36] B. Ali, , R. A. Gill, S. Yang, M. B. Gill, M. A. Farooq, D. Liu, & W. Zhou, "Regulation of cadmium-induced proteomic and metabolic changes by 5-aminolevulinic acid in leaves of Brassica napus L." PLoS One 10(4), e0123328 (2015).

[37] T. A. Tran, & L. P. Popova, "Functions and toxicity of cadmium in plants: recent advances and future prospects."*Turkish journal of Botany* 37(1), 1-13 (2013).

[38] A. Wahid, A. Ghani, & F. Javed, "Effect of cadmium on photosynthesis, nutrition and growth of mungbean" Agronomy for sustainable development 28(2), 273-280 (2008).

[39] S. A. Hasan, Q. Fariduddin, B. Ali, S. Hayat, & A. Ahmad, "Cadmium: toxicity and tolerance in plants" *J Environ Biol*" 30(2), 165-174 (2009).

[40]M. T. Rafiq, R. Aziz, X. Yang, W. Xiao, M. K. Rafiq, B. Ali, & T. Li, "Cadmium phytoavailability to rice (Oryza sativa L.) grown in representative Chinese soils. A model to improve soil environmental quality guidelines for food safety", Ecotoxicology and environmental safety 103, 101-107 (2014).

[41] A. Baryla, P. Carrier, F. Franck, C. Coulomb, C. Sahut, & M. Havaux, "Leaf chlorosis in oilseed rape plants (Brassica napus) grown on cadmium-polluted soil: causes and consequences for photosynthesis and growth" *Planta* 212(5), 696-709 (2001).

[42] M. Rodríguez-Serrano, M. C. Romero-Puertas, D. M. Pazmino, P. S. Testillano, M. C. Risueño, L. A. Del Río, & L. M. Sandalio, "Cellular response of pea plants to cadmium toxicity: cross talk between reactive oxygen species, nitric oxide, and calcium" *Plant Physiology* 150(1), 229-243 (2009).

[43]A. Małecka, A. Piechalak, I. Morkunas, & B. Tomaszewska, "Accumulation of lead in root cells of Pisum sativum" *Acta Physiologiae Plantarum* 30(5), 629-637(2008).

[44] E. A. Peroza, A. Al Kaabi, W. Meyer-Klaucke, G. Wellenreuther, & E. Freisinger, "The two distinctive metal ion binding domains of the wheat metallothionein Ec-1" *Journal of inorganic biochemistry* 103(3), 342-353 (2009).

[45] S. Singh, P. Parihar, R. Singh, V. P. Singh, & S. M. Prasad, "Heavy metal tolerance in plants: role of transcriptomics, proteomics, metabolomics, and ionomics" Frontiers in plant science 6, 1143 (2016).

[46] P. Ahmad, A. A. Abdel Latef, E. F. Abd_Allah, A. Hashem, M. Sarwat, N. A. Anjum, &S. Gucel, "Calcium and potassium supplementation enhanced growth, osmolyte secondary metabolite production, and enzymatic antioxidant machinery in cadmium-exposed chickpea (Cicer arietinum L.)" Frontiers in Plant Science 7, 513 (2016).

[47] C. Loix, M. Huybrechts, J. Vangronsveld, M. Gielen, E. Keunen, & A. Cuypers, "Reciprocal interactions between cadmium-induced cell wall responses and oxidative stress in plants" Frontiers in Plant Science, 8, 1867 (2017).

[48]B. A. Lajayer, M. Ghorbanpour, & S. Nikabadi, "Heavy metals in contaminated environment: destiny of secondary metabolite biosynthesis, oxidative status and phytoextraction in medicinal plants" Ecotoxicology and Environmental Safety 145, 377-390 (2017).

[49] F. T. Bingham, "Bioavailability of Cd to food crops in relation to heavy metal content of sludge-amended soil" Environmental Health Perspectives 28, 39-43 (1979).

[50] J. C. Van Loon, J. Lichwa, & D. Ruttan, "A study of the determination and distribution of cadmium in samples collected in a heavily industrialized and urbanized region (Metropolitan Toronto)" *International Journal of Environmental Analytical Chemistry* 3(2), 147-160 (1973).

[51]A. Ullah, S. Heng, M. F. H. Munis, S. Fahad, & X. Yang, "Phytoremediation of heavy metals assisted by plant growth promoting (PGP) bacteria: a review" Environmental and Experimental Botany 117, 28-40 (2015).

[52] Kathal Rekha, Priti Malhotra and Vidhi Chaudhary, "Phytoremediation of cadmium from polluted soil" *Journal of Bioremediation and Biodegradation* 7, no. 6 (2016).

[53]H. M. Chen, C. R. Zheng, Tu, C., & Shen, Z. G. "Chemical methods and phytoremediation of soil contaminated with heavy metals", *Chemosphere*, 41(1-2), 229-234 (2000).

[54] S. Lata, H. P. Kaur, &T. Mishra, "Cadmium bioremediation: a review" *Int J Pharm Sci Res*, 10(9), 4120-28 (2019).

[55] R. B. Meagher, "Phytoremediation of toxic elemental and organic pollutants" Current opinion in plant biology 3(2), 153-162 (2000).

[56] I. Kuiper, E. L. Lagendijk, G. V. Bloemberg, &B. J. Lugtenberg, "Rhizoremediation: a beneficial plant-microbe interaction" Molecular plant-microbe interactions 17(1), 6-15 (2004).

[57] Z. Chen, Y. Zhao, L. Fan, L. Xing, & Y. Yang, "Cadmium (Cd) localization in tissues of cotton (Gossypium hirsutum L.), and its phytoremediation potential for Cd-contaminated soils" *Bulletin of environmental contamination and toxicology* 95(6), 784-789 (2015).
[58] R. V. Khandare, & S. P. Govindwar, "Phytoremediation of textile dyes and effluents: Current scenario and future prospects" Biotechnology Advances 33(8), 1697-1714 (2015).

[59] A. Rashid, T. Mahmood, F. Mehmood, A. Khalid, B. Saba, A. Batool, & A. Riaz, "Phytoaccumulation, competitive adsorption and evaluation of chelators-metal interaction in lettuce plant" Environmental Engineering & Management Journal (EEMJ) 13(10) (2014).

[60] M. Kamal, A. E. Ghaly, N. Mahmoud, & R. Cote, "Phytoaccumulation of heavy metals by aquatic plants" Environment international 29(8), 1029-1039 (2004).

[61] R. Kathal, P. Malhotra, L. Kumar, & P. L. Uniyal, "Phytoextraction of Pb and Ni from the polluted soil by Brassica juncea L." *J Environ Anal Toxicol* 6(394), 2161-0525 (2016).

[62] C. A. Harguinteguy, M. L. Pignata, & A. Fernández-Cirelli, "Nickel, lead and zinc accumulation and performance in relation to their use in phytoremediation of macrophytes Myriophyllum aquaticum and Egeria densa" Ecological engineering 82, 512-516 (2015).

[63] R. Zurayk, B.Sukkariyah, R. Baalbaki, & D. Abi Ghanem, "Ni Phytoaccumulation in Mentha aquatica L. and Mentha sylvestris L." Water, Air, and Soil Pollution 139(1), 355-364 (2002).

[64] M. Casado, H. M. Anawar, A. Garcia-Sanchez, & I. S. Regina, "Cadmium and zinc in polluted mining soils and uptake by plants (El Losar mine, Spain)" *International Journal of Environment and Pollution* 33(2-3), 146-159 (2008).

[65] V. Mudgal, N.Madaan, A. Mudgal, R. B. Singh, & S. Mishra, "Effect of toxic metals on human health" *The open Nutraceuticals journal* 3(1) (2010).

[66] W. Liu, Q. Zhou, Z. Zhang, T. Hua, & Z. Cai, "Evaluation of cadmium phytoremediation potential in Chinese cabbage cultivars" *Journal of Agricultural and Food Chemistry* 59(15), 8324-8330 (2011).

[67] N. S.Bolan, D. C. Adriano, P. A. Mani, & A. Duraisamy, "Immobilization and phytoavailability of cadmium in variable charge soils. II. Effect of lime addition", *Plant and Soil* 251(2), 187-198 (**2003**).

[68] E. Lombi, F. J. Zhao, S. P. McGrath, S. D. Young, & G. A. Sacchi, "Physiological evidence for a high-affinity cadmium transporter highly expressed in a Thlaspi caerulescens ecotype" *New Phytologist* 149(1), 53-60 (2001).

[69] Y. T. Tang, R. L. Qiu, X. W. Zeng, R. R. Ying, F. M. Yu, & X. Y. Zhou, "Lead, zinc, cadmium hyperaccumulation and growth stimulation in Arabis paniculata Franch" Environmental and Experimental Botany 66(1), 126-134 (2009).

[70] V. Subhashini, & A. V. V. S. Swamy, "Phytoremediation of cadmium and chromium contaminated soils by Cyperus rotundus L."*Am Int J Res Sci Technol Eng Math*, 6, 97-101 (2014).

[71] G. Yu, P. Jiang, X. Fu, J. Liu, G. I. Sunahara, Z. Chen, & X. Wang, "Phytoextraction of cadmium-contaminated soil by Celosia argentea Linn.: A long-term field study" *Environmental Pollution* 266, 115408 (2020).

[72] C. Turgut, M. K. Pepe, & T. J. Cutright, "The effect of EDTA and citric acid on phytoremediation of Cd, Cr, and Ni from soil using Helianthus annuus" *Environmental pollution* 131(1), 147-154 (2004).

[73] M.Shahid, A. Austruy, G. Echevarria, M. Arshad, M. Sanaullah, M. Aslam, & C. Dumat, "EDTA-enhanced phytoremediation of heavy metals: a review" *Soil and Sediment Contamination: An International Journal* 23(4), 389-416 (2014).

[74] D. K. Bagga, & S. Peterson, "Phytoremediation of arsenic-contaminated soil as affected by the chelating agent CDTA and different levels of soil pH" Remediation Journal: The Journal of Environmental Cleanup Costs, Technologies & Techniques, 12(1), 77-85 (2001).

[75] A. Kasiulienė, & V. Paulauskas, In-situ phytoremediation: a review of natural and chemically assisted phytoextraction, In Annual 19th International Scientific Conference Proceedings (Vol. 2, p. 107) (2013).

[76] K. Wenger, S. K. Gupta, G. Furrer, & R. Schulin, "Zinc extraction potential of two common crop plants, Nicotiana tabacum and Zea mays" *Plant and Soil* 242(2), 217-225 (2002).

[77] I. Alkorta, J. Hernández-Allica, J. M. Becerril, I. Amezaga, I., Albizu, & C. Garbisu, "Recent findings on the phytoremediation of soils contaminated with environmentally toxic heavy metals and metalloids such as zinc, cadmium, lead, and arsenic" Reviews in Environmental Science and Biotechnology 3(1), 71-90 (2004).

[78] W. R. Berti, & S. D. Cunningham, "Phytostabilization of metals" In I. Raskin & B. D. Ensley (Eds.) Phytoremediation of toxic metals: Using plants to clean-up the environment (pp. 71–88). New York: Wiley (2000).

[79]F. F. Munshower, D. R. Neuman, & S. R. Jennings, "Phytostabilization Permanence Within Montana's Clark Fork River Basin Superfund Sites" In National Meeting of the American Society of Mining and Reclamation, Billings, MT (2003).

[80] M. O. Mendez, & R. M. Maier, "Phytostabilization of mine tailings in arid and semiarid environments—an emerging remediation technology" *Environmental health perspectives*, 116(3), 278-283 (2008).

[81] S. Yadav, & J. Srivastava, "Research Article Cadmium Phytoextraction and Induced Antioxidant Gene Response in Moringa oleifera Lam" (2017).

[82] H. .Zhang, & M. Reynolds, "Cadmium exposure in living organisms: A short review" *Science of the Total Environment*, 678, 761-767 (2019).

[83] M. Regvar, K. Vogel-Mikuš, N.Kugonič, B. Turk, &F Batič, "Vegetational and mycorrhizal successions at a metal polluted site: Indications for the direction of phytostabilisation?" *Environmental Pollution* 144(3), 976-984 (2006).

[84] D. A. Rouch, B. T. Lee, & A. P. Morby, "Understanding cellular responses to toxic agents: a model for mechanism-choice in bacterial metal resistance" *Journal of industrial microbiology and biotechnology* 14(2), 132-141 (1995).

[85] Pilon-Smith, E. Phytoremediation. Ann. Rev. Plant Biol. 56, 15 (2005).

[86] O. C. Udoka, EE. O. kanem, M. A. Harami, & A. Tafawa, "Phytoaccumulation potentials of Tamarindus indica" *Int J Innov Sci Res*, 11, 72-78 (2014).

[87] N.Gorinova, M. Nedkovska, E. Todorovska, L. Simova-Stoilova, Z. Stoyanova, K. Georgieva, &R. Herzig, "Improved phytoaccumulation of cadmium by genetically modified tobacco plants (Nicotiana tabacum L.). Physiological and biochemical response of the transformants to cadmium toxicity", *Environmental Pollution* 145(1), 161-170 (2007).

[88] T. Redjala, I. Zelko, T. Sterckeman, V. Legué, &A. Lux, "Relationship between root structure and root cadmium uptake in maize", *Environmental and Experimental Botany* 71(2), 241-248 (2011).

[89] C. D. Jadia, & M. H. Fulekar, "Phytoremediation: The application of vermicompost to remove zinc, cadmium, copper, nickel and lead by sunflower plant" *Environmental Engineering & Management Journal* (EEMJ) 7(5) (2008).

[90] N. P. Singh, &A. R. Santal, "Phytoremediation of heavy metals: the use of green approaches to clean the environment. In Phytoremediation (pp. 115-129)", Springer, Cham (2015).

[91] United States. Environmental Protection Agency. Office of Pesticide Programs. Assigning values to non-detected/non-quantified pesticide residues in human health food exposure assessments. Office of Pesticide Programs, US Environmental Protection Agency (2000).

[92] B. D. Ensley, "Rational for use of phytoremediation", Phytoremediation of toxic metals (2000).

[93] P. Mahajan, & J. Kaushal, "Role of phytoremediation in reducing cadmium toxicity in soil and water" Journal of toxicology, **(2018)**.

[94] S. Khaokaew, & G. Landrot, "A field-scale study of cadmium phytoremediation in a contaminated agricultural soil at Mae Sot District, Tak Province, Thailand:(1) Determination of Cd-hyperaccumulating plants" *Chemosphere* 138, 883-887 (2015).

[95] H. Li, X. Li, L. Xiang, H. M. Zhao, Y. W. Li, Q. Y. Cai, & M. H. Wong, "Phytoremediation of soil co-contaminated with Cd and BDE-209 using hyperaccumulator enhanced by AM fungi and surfactant" Science of the Total Environment 613, 447-455 (2018).

[96] F. N. Moreno, C. W. N. Anderson, R. B. Stewart, B. H. Robinson, R. Nomura, & M. Gomshei, "Mercury phytoextraction and phytovolatilisation from hg-contaminated artisanal mine sites" Phytoremediat Mercur Mine Wastes 2004a, 147-159 (2004).

[97] U. Song, & H. Park, "Importance of biomass management acts and policies after phytoremediation" *Journal of Ecology and Environment* 41(1), 1-6 (2017).

[98] B. Y. Zhang, J. S. Zheng, & R. G. Sharp, "Phytoremediation in engineered wetlands: mechanisms and applications" Procedia Environmental Sciences 2, 1315-1325 (2010).

[99] A. M. Shackira, & J. T. Puthur, "Enhanced phytostabilization of cadmium by a halophyte— Acanthus ilicifolius L" *International journal of phytoremediation* 19(4), 319-326 (2017).

[100] M. Varun, R. D'Souza, D. Kumar, & M. S. Paul, "Bioassay as monitoring system for lead phytoremediation through Crinum asiaticum L" Environmental monitoring and assessment 178(1), 373-381 (2011).

[101] J. A. Domínguez Núñez, R. Planelles González, J. A. Rodríguez Barreal, & J. A. Saiz de Omeñaca González, "Influence of water-stress acclimation and Tuber melanosporum mycorrhization on Quercus ilex seedlings" *Agroforestry systems* 75(3), 251-259 (2009).

[102]H. Ali, E. Khan, , &M. A. Sajad, "Phytoremediation of heavy metals—concepts and applications" *Chemosphere* 91(7), 869-881 (2013).

[103] S. Khalid, M. Shahid, N. K. Niazi, B. Murtaza, I. Bibi, & C. Dumat, "A comparison of technologies for remediation of heavy metal contaminated soils" *Journal of Geochemical Exploration* 182, 247-268 (2017).

[104] K. R. Kim, & G. Owens, "Potential for enhanced phytoremediation of landfills using biosolids–a review" *Journal of environmental management* 91(4), 791-797 (2010).